

APIGENIN 5-GLUCOSIDE FROM *Equisetum arvense*

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UDC 547.972

The glycosidation of flavones at the 5-OH group is rarely found in nature [1, 2]. The finding in *Equisetum arvense* L. of luteolin 5-glucoside has been reported previously [5], but later the question of the finding of this compound was placed under doubt [1, 4].

Continuing a study of the flavonoids of the epigeal part of *E. arvense* L. (field horsetail) we have isolated from an ethyl acetate extract by chromatography on a polyamide sorbent a white crystalline substance with mp 255-256°C,  $[\alpha]^{20} -22.7^\circ$  (c 0.175; methanol)  $\lambda_{\text{max}}^{\text{(CH}_3\text{OH)}}$  262, 333 (log  $\epsilon$ , 413; 416);  $\lambda_{\text{max}}^{\text{(AlCl}_3\text{)}}$  262, 335;  $\lambda_{\text{max}}^{\text{(AlCl}_3\text{+HCl)}}$  263, 335;  $\lambda_{\text{max}}^{\text{(CH}_3\text{COONa)}}$  271, 305, 370;  $\lambda_{\text{max}}^{\text{(CH}_3\text{ONa)}}$  270, 330, 390 nm. The band of the valence vibrations of a carbonyl group is present at 1630  $\text{cm}^{-1}$  and that of hydroxy groups in the 3300-3400- $\text{cm}^{-1}$  region.

From the results of quantitative acid and enzymatic hydrolytic cleavage it can be seen that the substance under investigation is a monoside containing D-glucose as the carbohydrate component and apigenin as the aglycone [5].

The compound under consideration possesses a bright-blue fluorescence in UV light, which is characteristic for flavone 5-glycosides [1]. The absence of a bathochromic shift in the presence of  $\text{AlCl}_3$  for the glycoside, and also the absence of the signal of the proton of the 5-OH group in the weak field of the PMR spectrum (with dimethyl sulfoxide as solvent, B 487B radiospectrometer) in comparison with apigenin shows that the glucose is attached at position 5.

The capacity of the glycoside for being split by  $\beta$ -glucosidase shows the  $\beta$ -configuration of the glycosidic bond. On comparing the molecular rotations of the glycoside with the corresponding figures for phenyl glycosides it can be seen that the carbohydrate substituent is  $\beta$ -D-glucopyranose. The pyranose form is also confirmed by the results of IR spectroscopy (1100, 1070, and 1030  $\text{cm}^{-1}$ ) [16]. A doublet at  $\delta$  5.02 ppm (1H, J = 6 Hz) corresponds to the anomeric proton of the  $\beta$ -glucose in position 5 [7].

Thus, the compound under investigation has the structure of apigenin 5-O- $\beta$ -D-glucopyranoside.

Apigenin 5-glucoside has been isolated previously from *Amorpha fruticosa* [8] and was later synthesized, but its physicochemical constants, with the exception of the melting point, were not given. The melting point of the apigenin 5-glucoside that we have identified (255-256°C) differs from that given in the literature (295°C) [8, 9].

This is the first time that apigenin 5-O- $\beta$ -D-glucopyranoside has been found in the family Equisetaceae.

LITERATURE CITED

1. J. B. Harborn, *Phytochemistry*, **6**, 1569 (1967).
2. V. A. Bandyukova and É. T. Avanesov, *Khim. Prirodn. Soedin.*, 413 (1972).
3. H. Nakamura and G. Hukuti, *J. Pharm. Soc. Jap.*, **60**, 449 (1940).
4. N. A. M. Salleh, W. Majak, and G. H. N. Towers, *Phytochemistry*, **11**, 1095 (1973).

Irkutsk Institute of Organic Chemistry, Siberian Branch, Academy of Sciences of the USSR. Translated from *Khimiya Prirodnikh Soedinenii*, No. 5, pp. 666-667, September-October, 1974. Original article submitted April 30, 1974.

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5. A. I. Syrchina, M. G. Voronkov, and N. A. Tyukavkina, *Khim. Prirodn. Soedin.*, 671 (1973).
6. I. L. Kovalev and V. I. Litvinenko, *Khim. Prirodn. Soedin.*, 233 (1965).
7. T. J. Mabry, K. R. Markham, and M. B. Thomas, *The Systematic Identification of Flavonoids*, Springer, New York (1970), p. 269.
8. R. Goto and M. Taki, *J. Pharm. Soc. Jap.*, 58, 933 (1938).
9. G. Zemplen and L. Mester, *Ber.*, 76, 776 (1943).